



Certification Boulevard

Test Your Knowledge of Miscellaneous Technical Topics

1. What does the "M" represent in the process parameter F/M ratio?
 - a. lbs volatile microorganisms in aeration**
 - b. gpd plant flow
 - c. lbs/day influent CBOD₅ entering aeration
 - d. gpd waste sludge leaving aeration
2. What may be the cause if CBOD₅ removal in a primary clarifier is less than 15%?
 - a. The plant is underloaded
 - b. The detention time is too short**
 - c. The detention time is too long
 - d. 15% CBOD₅ removal from a primary clarifier is greater than expected
3. What is a typical range for gas production in a properly operated anaerobic digestion process?
 - a. 1 to 2 ft³ per lb VS reduced
 - b. 5 to 7 ft³ per lb VS reduced
 - c. 11 to 20 ft³ per lb VS reduced**
 - d. 40 to 60 ft³ per lb VS reduced
4. What is created when chlorine reacts with ammonia in the effluent stream?
 - a. Chloramines
 - b. Free residual
 - c. Mono residual
 - d. Combined residual
 - e. Both "a & d"**
5. Which group of bacteria are responsible for converting NO₂ to NO₃?
 - a. Heterotrophic
 - b. Nitrosomonas
 - c. Nitrobacter**
 - d. Fermenters
6. Given the following data, what is the sludge blanket detention time in a secondary clarifier?
 - 50 Foot Diameter
 - 3 Foot Sludge Blanket Depth
 - 2:12 Floor Slope
 - 0.5 mgd RAS Rate
 - a. 2.6 days
 - b. 3.1 hours**
 - c. 2.1 hours
 - d. 1.3 days

Sludge Blanket Detention Time, hours

$$\begin{aligned} &= \frac{\text{Total Gallons of Sludge in Clarifier}}{\text{RAS Rate, gpd}} \times 24 \text{ hrs/day} \\ \text{Gals Sludge in Blanket} &= \pi r^2 \times \text{blanket depth} \times 7.48 \\ &= 3.14 \times 25 \text{ ft} \times 25 \text{ ft} \times 3 \text{ ft.} \times 7.48 \text{ gal/ft}^3 \\ &= 44,038 \text{ gals} \\ \text{Gals Sludge in Cone} &= \frac{1}{3} \pi r^2 \times \text{cone depth} \times 7.48 \\ &= (3.14 \times 25 \text{ ft} \times 25 \text{ ft} \times (50 \text{ ft} \div 12 \text{ in/ft}) \times 7.48 \text{ gal/ft}^3) \div 3 \\ &= 20,388 \text{ gals} \\ &= \frac{44,038 \text{ gals in blanket} + 20,388 \text{ gals in cone}}{500,000 \text{ gpd RAS}} \times 24 \text{ hrs/day} \\ &= 3.09 \text{ hours} \end{aligned}$$

7. Given the following data, what is the cost of polymer used, in dollars per dry ton of sludge processed, by a Belt Filter Press?

- Total sludge feed is 144,250 gpd
- Feed sludge concentration is 2.75% TS
- Total neat polymer used is 35 gpd
- Polymer specific gravity (S.G.) is 1.03
- Polymer cost is \$0.69 per pound

- a. \$45.24 per dt
- b. \$12.54 per dt**
- c. \$37.64 per dt
- d. \$12.18 per dt

Cost of Polymer per Dry Ton of Sludge Processed

$$\begin{aligned} &= \frac{\text{Total Cost of Polymer Utilization}}{\text{Total Dry Tons Sludge Processed}} \\ \text{Total Cost of Polymer Used} &= \text{gpd neat polymer used} \times \text{weight per gal} \times \$ \text{ per pound} \\ &= 35 \text{ gpd} \times (8.34 \text{ lbs/gal} \times 1.03 \text{ S.G.}) \times \$0.69 \text{ per lb} \\ &= \$207.45 \text{ per day polymer used} \\ \text{Total Dry Tons Sludge} &= \text{lbs/day sludge feed} \div 2,000 \text{ lbs/ton} \\ &= (0.14425 \text{ mgd} \times 27,500 \text{ mg/l} \times 8.34 \text{ lbs/gal}) \div 2,000 \\ &= 16.54 \text{ dtpd (dry tons per day)} \\ \text{Cost per Dry Ton} &= \frac{\$207.45 \text{ per day}}{16.54 \text{ dtpd}} \\ &= \$12.54 \text{ polymer per dry ton sludge processed} \end{aligned}$$

8. Given the following data, what is the Specific Oxygen Utilization Rate (SOUR) in an aerobic digester?

- OUR test starting D.O. is 7.2 mg/l
- OUR test ending D.O. is 4.0 mg/l
- OUR test time is 10 minutes
- Digested sludge VSS concentration is 14,500 mg/l
- Digested sludge volatile fraction is 70%

a. 1.32 mg/hr/gm

b. 0.78 mg/hr/gm

c. 1.6 mg/hr/gm

d. 0.93 mg/hr/gm

$$\text{SOUR, mg/hr/gm TS} = \frac{\text{OUR, mg/l/hr}}{\text{TS, gm/l}}$$

$$\text{OUR, mg/l/hr} = \frac{\text{Starting D.O., mg/l} - \text{Ending D.O., mg/l}}{\text{Test Time, mins}} \times 60 \text{ mins/hr}$$

$$= \frac{7.2 \text{ mg/l} - 4.0 \text{ mg/l}}{10 \text{ mins}} \times 60 \text{ mins/hr}$$

$$= 19.2 \text{ mg/l/hr}$$

$$\text{TS, gm/l} = (\text{VSS, mg/l} \div \text{Volatile Fraction, \%}) \div 1,000 \text{ mg/gm}$$

$$= (14,500 \text{ mg/l} \div 0.7) \div 1,000 \text{ mg/gm}$$

$$= 20.71 \text{ gm TS}$$

$$\text{SOUR, mg/hr/gm} = 0.93 \text{ mg/hr/gm}$$

9. Which effluent quality condition may cause the most efficiency problems with a UV disinfection process?

a. High NH_3

b. High NO_3

c. High NO_2

d. High TSS

10. What is the term when $\text{NH}_3\text{-N}$ and Org-N are added together?

a. TN

b. SON

c. NO_3

d. TKN

e. NO_2

f. NO_x

g. None of the above